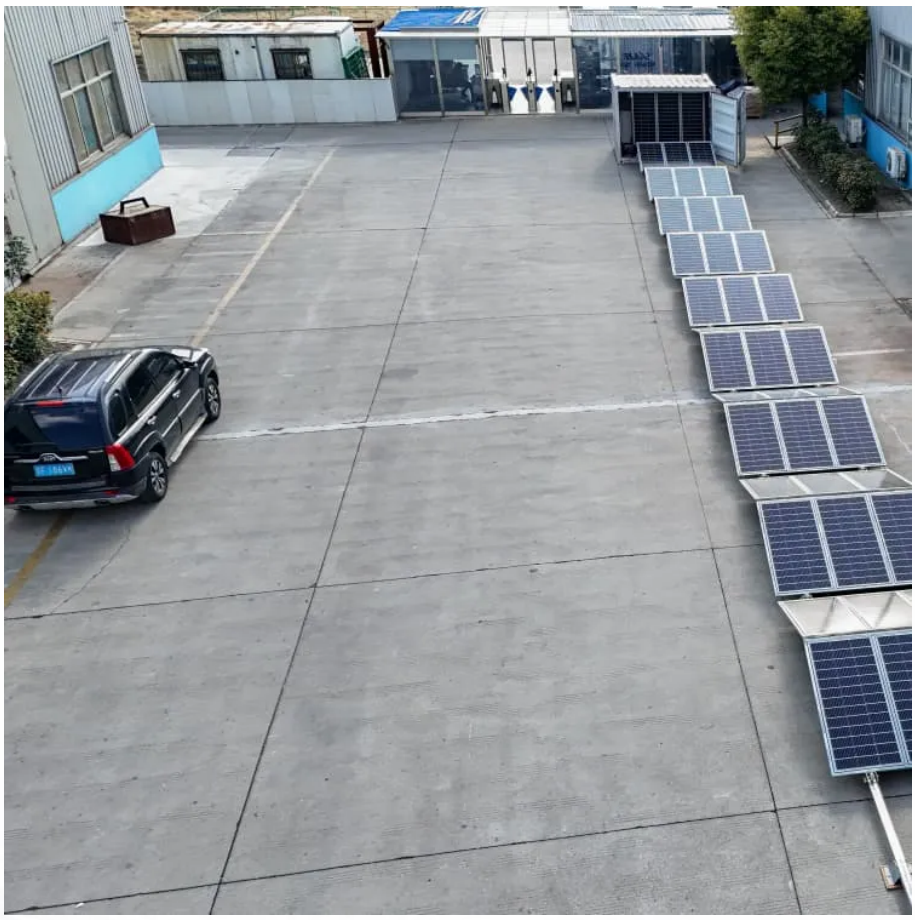
WALMER ENERGY

Electromotive force of flow battery



Overview

What is electromotive force in a battery?

So there has to be other force there that push them. Such force per unit charge is usually called electromotive force, but that term is too general. It is more descriptive to call it "chemical electromotive forces", because they arise as a result of chemical reactions in the battery. There are other kinds of electromotive forces.

What is electromotive force (EMF)?

Electromotive Force (EMF) is the work done by the per unit charge while moving from the positive end to the negative end of the battery. It can also be defined as the energy gain per unit charge while moving from the positive end to the negative end of the battery.

Why is the electromotive force of a battery negative?

During charging, the flow of the current in the circuit is opposite to the normal flow of the current. Thus, the Electromotive Force of a battery can easily be negative when the battery charges.

How does electromotive force work in a cell?

The electric charges that flow around the circuit transfer energy from the source to the device. In a cell, chemical reactions convert chemical energy into electrical energy. This energy pushes the free electrons to move from the negative terminal to the positive terminal of the cell. What do you mean by electromotive force?

Electromotive force of flow battery

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